

Unusual Bioactive Annonaceous Acetogenins from Goniothalamus giganteus

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Abstract- Pyranicin (1) and pyragonicin (2) are the first mono-tetrahydropyran annonaceous acetogenins, and goniotrionin (3) possesses an unusual hydroxylated-allylic moiety. 1-3 were isolated from the bark of Goniothalamus giganteus using activity-directed fractionation with the brine shrimp lethality test. Both 1 and 2 are selectively cytotoxic against the pancreatic cell line (PACA-2) in a panel of six human solid tumor cell lines with 1 showing ten times the potency of adriamycin, while 3 showed more potent selectivity against the breast cell line (MCF-7).

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INTRODUCTION

Annonaceous acetogenins are a relatively new class of natural polyketides which have promising anticancer, antiinfective, and pesticidal properties. Structurally, most of these longchain fatty acid derivatives may be classified into three major groups, i.e., mono-tetrahydrofuran (THF), adjacent bis-THF, and nonadjacent bis-THF subclasses. Only two, of approximately 250 previously reported Annonaceous acetogenins, have a tetrahydropyran (THP) ring, and in these compounds the THP rings are either adjacent to or non-adjacent to a THF ring; both of these previously known THP bearing compounds have been isolated recently from Rollinia mucosa (Jaca.) Baill. (Annonaceae). 23 Goniothalamus giganteus Hook. f. & Thomas (Annonaceae) is a tree native to Thailand; in our further bioactivity-directed search of its bark for antitumor compounds, 4 guided by lethality to brine shrimp larvae (BST), 5 we have now isolated the first mono-THP annonaceous acetogenins, pyranicin (1) and pyragonicin (2) (Figure 1); also isolated was goniotrionin (3, Figure 1), a highly cytotoxic mono-tetrahydrofuran with an unusual hydroxylated-allylic moiety pseudo-threo to the ring flanked-hydroxyl. Both 1 and 2 also represent the first C₃₅ THP bearing acetogenins, and this finding adds a new structural type to this family of natural compounds. 1 was about ten times as potent as 2 in cytotoxicity against a panel of six human tumor cell lines, and both showed selectivities toward the pancreatic cell line (PACA-2) with potencies equal to or ten times as potent as adriamycin. Compound 3 was about 10⁵ times more potent than adriamycin against the breast cancer cell line (MCF-7).

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RESULTS AND DISCUSSION

The molecular formulae of 1 and 2 were both determined as $C_{35}H_{64}O_7$ by HRCIMS (found 597.4711, calcd. 597.4730; found 597.4741, calcd. 597.4730, respectively), apparently suggesting that they are C_{35} acetogenins with four hydroxyls (Figure 1). The retention time of 2 on normal phase HPLC was a little longer than that of 1, suggesting a more polar isomer. Although the ¹H NMR spectra of 1 (Table 1) and 2 (Table 2) showed peaks diagnostic of the 2,4-disubstituted α , β -unsaturated γ -lactone terminal with a 4-OH¹ and a hydroxylated THP ring like that of mucocin,² they also presented certain salient features. For 1 and 2, the absence of signals of a THF ring, the coupling pattern of H-19(H-17) and H-20(H-18), and the chemical shifts of H/C-19(H/C-17) suggested mono-THP ring acetogenins with a different stereochemistry than that of mucocin.²

Figure 1. The chemical structures of 1, 1a, 1b, 2, 2a, 2b, and 3.

The skeletal structures of 1 and 2 were established by ¹H-¹H COSY, NOESY, and EIMS and by comparing the ¹H and ¹³C NMR data with those of mucocin² and known acetogenins. ¹ The presence of a hydroxylated THP moiety with one flanking hydroxyl in both 1 and 2 was evident from the degree of unsaturation, the single relayed COSY cross peaks between H-16(14)/18(16), an intensive NOESY cross peak between H-16(14)/20(18), the lack of the same cross peak in its regular and single relayed COSY spectra, and COSY cross peaks at δ 3.46(3.50)/3.19(3.24) (later assigned as H-15(13)/16(14) in 1 and 2, respectively). The presence of four hydroxyls in both 1 and 2 was suggested by the successive losses of four H₂O molecules (m/z, 18) from the [MH]⁺ in the CIMS. The position of the hydroxylated-THP ring, the flanking hydroxyl, and the other hydroxyls along the hydrocarbon chain were determined by EIMS and HREIMS (Figure 2), and 1

and 2 are identical except for the placement of the THP ring and its flanking hydroxyl beginning at C-15 in 1 and at C-13 in 2.

Table 1. NMR Spectral Data (δ) for 1, 1a, and 1b.

Table 1. NMR Spectral Data (o) for 1, 1a, and 1b.								
proton	¹³ C ^a	¹ H NMR (<i>J</i> in Hz)						
carbon	1	1	la	1b	451b-1c			
1	174.7	-						
2	131.1	-						
3b		2.41 ddt (15, 8.5, 1.5)	2.54	2.58	-0.04			
3a	33.3	2.52 ddt (15.0, 3.5, 1.5)	2.56	2.66	-0.1			
4	69.8	3.84 m	5.31	5.35	$R^{\mathbf{b}}$			
5	37.3	1.49 m	1.66	1.57	+0.09			
6-8	25.3-	1.18-1.71						
9	37.2	1.45 m						
10	71.7	3.60 m	5.02	5.00				
11	37.2	1.45 m						
12-13	25.3-	1.18-1.71						
14	31.5°	1.41 m, 1.49 m	1.60	1.48	+0.12			
15	73.9	3.46 dt (7.5, 3.0)	5.02	5.02	R^{b}			
16	81.1	3.19 ddd (10.5, 7.0, 2.5)	3.45	3.52	-0.07			
17	21.5	1.45 m, 1.59 m	1.40-1.50	1.35, 1.72				
18	30.5	1.68 m, 2.01 m	1.73, 2.08	1.73, 2.08	0, 0			
19	66.1	3.61 m	5.00	5.03	S⁴			
20	80.0	3.34 ddd (7.5, 6.0, <1.0)	3.42	-0.07				
21	25.3-	1.47 m, 1.62 m	1.36, 1.42	-0.19, -				
22	32.3°	1.18-1.71 m						
23-29	25.3-	1.18-1.71 m						
30	31.9°	1.18-1.71 m						
31	22.6	1.30 m						
32	14.1	0.88 t (7.0)						
33	151.9	7.19 q (1.5)	6.72	6.95	-0.23			
34	78 .0	5.06 qq (7.0, 1.5)	4.86	4.92	-0.06			
35	19.0	1.44 d (7.0)	1.29	1.30	-0.01			

assignment assisted by HMQC and HMBC.

A threo relationship at C-15(13)/16(14) of 1 and 2 was suggested by extending Born's rule and by comparing ¹H- and ¹³C-NMR chemical shifts with those of mucocin^{2,6}. Born's rule had predicted a threo relative stereochemistry at the equivalent positions of C-19/C-20 in mucocin, which was then secured by applying Mosher ester methodology to its formaldehyde acetal derivative. The cis stereochemistry (referring to the side chains at C-16(14) and C-20(18)) was assigned to the THP ring because an intense cross peak at H-16(14)/H-20(18) was

^b absolute configuration of carbinol center.

^c signals are interchangeable.

^d assignment assisted by 2D NOESY.

observed in the NOESY spectra (Figure 3); the two side chains at C-20(18) and C-16(14) should both assume the equatorial positions which are energetically favorable.

Table 2. NMR Spectral Data (δ) for 2, 2a, and 2b

Table 2. NMK Spectral Data (6) for 2, 2a, and 2b.								
proto	¹³ C ^a	1 H NMR (J in Hz)						
carbo	2	2	2a	2b	^{∆8} 2b-2c			
1	174.5	-						
2	131.1	-						
3 b		2.40 ddt (15, 8.5, 1.5)	2.54	2.57	-0.03			
3 a	33.0	2.53 ddt (15.0, 3.5, 1.5)	2.60	2.67	-0.07			
4	69.8	3.85 m	5.30	5.34	R^b			
5	37.3	1.48 m	1.61	1.56	+0.05			
6-8	25.3-	1.18-1.71						
9	37.0	1.45 m						
10	71.5	3.63 m	4.99	5.02	R^{b}			
11	25.3-	1.51 m, 1.68 m						
12	25.3-	1.48 m, 1.69 m						
13	74.4	3.50 dt (7.5, 3.0)	5.02	4.99	R^b			
14	80.9	3.24 ddd (10.5, 7.0, 2.5)	3.38	3.48	-0.1			
15	21.3	1.48 m, 1.58 m	1.32, 1.40	1.32, 1.26	0, +0.14			
1 6	30.4	1.69 m, 2.00 m	1.58, 2.05	1.70, 2.07	-0.12, -			
17	66.0	3.63 m	4.99	5. 02	S°			
18	79.8	3.36 ddd (8.0, 5.5, <1.0)	3.32	3.40	-0.08			
19	31.4	1.49 m, 1.61 m	1.14, 1.25	1.35, 1.42	-0.18, -			
20-29	25.3-	1.18-1.71						
30	31.9	1.18-1.71 m						
31	22.6	1.30 m						
32	14.0	0.88 t (7.5)						
33	151.7	7.19 q (1.5)	6.73	6.95	-0.22			
34	77.9	5.06 qq (7.0, 1.5)	4.86	4.91	-0.05			
35	19.0	1.44 d (7.0)	1.27	1.30	-0.03			

^a assignment by HMQC and HMBC.

The OH group on the THP ring seems to assume the axial position. In mucocin² the THP hydroxyl group assumes the equatorial position showing a 1 H-NMR signal for H-23 at δ 3.28 and a 13 C-NMR signal for C-23 at δ 70.5, while H-19(17) in 1 and 2 shows 1 H-NMR signals at δ 3.61(3.63) and 13 C-NMR at δ 66.1(66.0), respectively; these large differences in values may indicate a different stereochemical environment within the THP ring system. Also, H-24 in mucocin² appears as a doublet of a triplet, while H-20(18) in 1 and 2 is ddd (with one J value <1 Hz) indicative of an e-a arrangement between H-19(17)/20(18); an e-e arrangement would require both large side chains to be in the axial positions which is highly unlikely. In addition, H-19(17)/20(18) gave very weak coupling in the COSY spectra which confirms the a-e spatial configuration between these two protons.

^b absolute configuration of carbinol center.

[&]quot; assignment assisted by 2D NOESY.

Figure 2. Diagnostic EIMS fragmentions of 1 and 2; ions in parentheses were not observed; * ions confirmed by HREIMS.

Figure 3. 2D-NOESY correlations of the hydroxylated-THP ring in both 1 and 2.

The absolute stereochemistries of 1 and 2 were determined by advanced Mosher ester methodology.⁷ The (R)- and (S)- tetra-MTPA esters of 1 and 2 were prepared, their ¹H NMR signals were assigned by the COSY spectra, and the corresponding $\Delta\delta$ (S-R) values were calculated (Tables 1 and 2). The negative value of -0.07(-0.1) at H-16(14) in both 1 and 2, respectively, suggested an R configuration at H-15(13) and, consequently, considering their relative stereochemistries, led us to assign R configurations to positions 16(14) and 20(18) and to assign S to C-19(17). Using Hoye's models⁸ for 4-OH 2,4-disubstituted γ -lactones, R and S configurations were assigned, respectively, to H-4 and H-34 in both 1 and 2.

Table 3. NMR Spectral Data for Goniotrionin (3).

position H/C	trai Data for Goniotrionin (3). H NMR (J in Hz)	¹³ C NMR (δ) ^a
1	-	174.6
2	-	131.2
3 a	2.53 ddt (15.5, 3.5, 1.0)	
3b	2.2.40 ddt (15, 8.0, 1.0)	33.4
4	3.85 m	69.9
4 5	1.48 m	37.2
6-8	1.18-1.65 m	25.5-37.2
9	1.53-1.64 m	35.4
10	3.89 m	79.3
11	2.03, 1.54 m	32.4
12	1.96, 1.60 m	28.2
13	3.85 m	81.8
14	3.73 dt (11.5, 3.5)	71.5
15	1.58 m	39.6
16	4.78 dt (8.0, 4.0)	65.0
17	5.48 m	131.6
18	5.44 m	132.2
19	2.10, 2.05 m	25.5-37.2
20-29	1.18-1.65 m	25.5-37.2
30	1.18-1.65 m	31.9
31	1.30 m	22.7
32	0.88 t (7.0)	14.1
33	7.19 q (1.5)	151.9
34	5.07 qq (7.0, 1.5)	78 .0
35	1.44 d (7.0)	19.1

assignment assisted by HMQC.

The absolute stereochemistry at H-10 in both 1 and 2 could not be assigned immediately from the COSY spectra of the (S)- and (R)-MTPA derivatives due to overlapping signals. This was resolved by directly comparing the R and S values at H-10 of both 1 and 2 to those of the per-MPTA derivatives of longicoricin⁹ (H-10 and H-15 diol) and goniothalamicin¹⁰ (H-10 and H-13 diol); consequently, the R configuration was assigned at H-10 for both 1 and 2.

Compound 3 was also isolated as a whitish wax. Its molecular weight was suggested by a molecular ion peak at m/z 579 [MH]⁺ in the FABMS. The HRFABMS gave m/z 579.4597 for the [MH]⁺ ion (calcd. 579.4625) corresponding to the molecular formula $C_{35}H_{62}O_{6}$.

Compound 3 showed an IR carbonyl absorption at 1740 cm⁻¹, a UV (MeOH) λ_{max} at 218 nm (log ϵ , 3.46), the proton resonances at δ 7.19, 5.07, 3.85, 2.54, 2.54, 2.54, 2.41, and 1.44 (Table 3), and carbon resonances at δ 174.6, 151.9, 131.2, 78.0, 69.9, and 19.1 (Table 3) all of which provided characteristic spectral features for an α,β -unsaturated γ -lactone fragment with a 4-OH.

Figure 4. Diagnostic EIMS fragments ions of goniotrionin (3)

The presence of three OH groups in 3 was suggested by a prominent OH absorption at 3368 cm⁻¹ in the IR spectrum and was confirmed by three successive losses of H_2O (m/z 18) from the [MH]⁺ in the CIMS and FABMS (Figure 4). The ¹³C NMR of 3 showed three carbon resonances due to oxygen-bearing carbons at δ 71.5 (C-14), 69.9 (C-4), and 65.0 (C-16) indicating the existence of three secondary OH moieties. The existence of a mono-THF ring with one flanking hydroxyl was suggested by the 2D-COSY cross peak, between δ 3.74 (H-14) and 3.85 (H-13), and by the carbon signals at δ 79.3 (C-10), 81.2 (C-13) and 71.5 (C-14). The unusual allylic moiety was identified by the 2D-cross peaks between δ 4.78 (H-16) and 5.44 (H-17), by the ¹H NMR signals at δ 5.44 (H-17), 5.48 (H-18), and 4.78 (H-16), and by the ¹³C NMR peaks at δ 65.0 (C-16), 131.6 (C-18), and 132.2 (C-17). The coupling pattern at δ 5.44/5.48 was quite complex. Nevertheless, the double bond was identified as being cis since a trans hydroxylated-allylic moiety in a long chain was found, by our group, ¹¹ to have a much larger chemical shift difference between the double bond protons (Δ =0.18 ppm) and carbons (Δ =5.0 ppm), while in 3 the difference was 0.04 between H-17 and H-18 and 0.6 between C-17 and C-18.

Table 4. Comparative chemical shift differences for 1,3 diols.

				(500 MHz	,δin ppm)			
	1,3 pseudo-erythro diol				1,3 pscudo-threo diol			
Compounds	δ ¹ H	δ ¹³ C	δ¹H	δ ¹³ C	δ ¹H	δ ¹³ C	δ ¹H	δ ¹³ C
Muricatocin A ^a	H-10	C-10	H-12	C-12				
	(3.94)	(72.8)	(3.86)	(72.6)				
Muricatocin Ca		, ,	` '	` /	H-10	C-10	H-12	C-12
					(3.94)	(69.6)	(3.86)	(69.2)
Isolated hydroxyl					(=)	()	()	(,
oxymethine ^b				¹ H (3.5 8)/ ¹³ C (72.2)				
Δδ	+0.36	+0.60	+0.28	+0.40	+0.36	-2.6	+0.28	-3.0
Goniotrionin (3)								
δ at H-14/C-14		¹ H (3.73)/ ¹³ C (71.5)						
Fujimoto model				` ′	` ,			
equivalent δ	¹ H (3.37)/ ¹³ C (74.2)							
Δδ (Η/С)		(+0.36/-2.7)						

^a The relative stereochemistries of the 1,3 diol were resolved by preparing acetonide derivatives. ^{14,15}

b Data taken from reticulatamol (a non-THF acetogenin bearing only one isolated hydroxyl at C-15 in a long chain hydroxarbon).

The relative stereochemistries of the ring system were established as *trans/threo* across C-10/C-13 and C-13/C-14, respectively, by comparing the ¹H NMR and ¹³C NMR to model compounds synthesized by Fujimoto *et al.*¹². The chemical shifts of H-14/C-14 were deviated from the equivalent signals in the model compound due to the effect of the 1,3 diol. A pseudo-*threo* spatial relationship between H-14 and H-16 was suggested by analyzing the ¹H and ¹³C chemical shifts values of 1,3 pseudo-*threo* and *-erythro* diol acetogenins (Table 4). In comparison with an isolated hydroxyl oxymethine along a hydrocarbon chain, hydroxyl oxymethines of 1,3 pseudo-*threo* diol usually experience ~+0.32 ppm upfield chemical shift in the ¹H-NMR and ~-2.8 ppm downfield chemical shift in the ¹³C-NMR. ^{1,13} Examples are muricatocins A and C with their H-10/H-12 diols. ^{14,15} The relative stereochemistries of the 1,3 diols in muricatocins A and C were previously confirmed as pseudo-*erythro* and *-threo*, respectively, by preparing the acetonide derivatives. ^{14,15} The chemical shift difference between H-14/C-14 in 3, from the equivalent signals in Fujimoto's model compounds, ¹² suggested a pseudo-*threo* relationship between H-14 and H-16 (Table 4).

Compound 3 is the first acetogenin with a hydroxylated-allylic moiety one carbon away from the THF ring system. The placement of the mono-THF, the allylic hydroxyl, and the other two hydroxyl groups were established based on careful EIMS spectra analysis of 3 (Figure 4) and on 2D single- and double relayed-COSY.

Table 5. Biological data for 1-3.

	BST ^a	YFM ^b	Cytotoxicity (ED ₅₀ , µg/mL)					
Compound	LC ₅₀	LC ₅₀	A-549°	MCF-7 ^d	HT-29°	A-498 ^f	PC-3 ⁸	PACA-2 ^h
	(µg/mL	(µg/mL)						
1	0.3	107.9	2.8×10 ⁻¹	3.9×10^{-1}	1.2	1.8×10 ⁻¹	4.1×10 ⁻¹	1.3×10^{-3}
2	0.9	73.8	2.0	1.6	2.8	1.3	1.2	5.8×10 ⁻²
3	NT	NT	7.7×10 ⁻³	5.3×10 ⁻⁶	3.4×10 ⁻¹	2.0×10 ⁻³	3.6×10 ⁻¹	5.4×10 ⁻³
rotenone	NT	0.8	NT	NT	NT	NT	NT	NT
ad riamycin ⁱ	NT	NT	7.8×10 ⁻³	1.2×10 ⁻¹	3.9×10 ⁻²	6.8×10 ⁻²	3.6×10 ⁻¹	1.6×10 ⁻²

^{*}Brine shrimp lethality test; *Yellow fever mosquito larvae test; *Human lung carcinoma;

⁶Human breast carcinoma; ⁶Human colon adenocarcinoma; ⁶Human kidney carcinoma;

Human prostate adenocarcinoma; human pancreatic carcinoma,

jiPositive control standards. NT: not tested.

Both 1 and 2 were quite active in the BST assay, 5 marginally active in the yellow fever mosquito larvae microtiter assay, 16 and selectively inhibitory against the PACA-2 cell line (pancreatic carcinoma) in a panel of six human solid tumor cell lines 17 (Table 3); 1 showed about ten times the potency of adriamycin in PACA-2 and is generally about ten times as potent as 2; this is consistant with other acetogenins in which the optimium structure activity relationship has these ring systems beginning at C-15. 18 Compound 3 was significantly cytotoxic against the panel of six cell lines with potent activity, 105 times that of adriamycin, against the breast cancer cell line (MCF-7).

Annonaceous acetogenins inhibit cancerous cells by the blockage of mitochondrial complex I (NADH-ubiquinone oxidoreductase)¹⁹ and also through the inhibition of the NADH oxidase prevalent in the plasma membranes of tumor cells.²⁰ These mechanisms deplete ATP and likely induce apoptosis (programmed cell death);²¹ pesticide-resistant German cockroaches and multidrug resistant tumor cells are especially thwarted by these actions probably through the inhibition of ATP-dependent efflux pumps.^{22,23}

EXPERIMENTAL SECTION

Instrumentation. Optical rotations were determined on a Perkin 241 polarimeter. IR spectra (film) were measured on a Perkin-Elmer 1600 FTIR spectrometer. UV spectra were taken in MeOH on a Beckman DU 640 series spectrophotometer. ¹H NMR, ¹H-¹H COSY, and ¹³C NMR spectra were obtained on a Varian VXR-500S spectrometer. Low resolution MS data were collected on a Finnigan 4000 spectrometer. High resolution CIMS were performed on a Kratos MS50. HPLC separations were performed with a Rainin Dynamax solvent delivery system (model SD-200) using a Dynamax software system and a silica gel column (Dynamax 60-A 250 x 21 mm) equipped with a Dynamax absorbance detector (model UV-1) set at 225 nm. Analytical TLC was carried out on silica gel plates (0.25 mm) developed with CHCl₃-MeOH (9:1) and visualized with 5% phosphomolybdic acid in EtOH.

Plant material. The stem bark of Goniothalamus giganteus (B-826538, PR-50604) was collected in Thailand in September 1978 under the auspices of Dr. Robert E. Perdue, Medicinal Plant Laboratory, USDA, Beltsville, MD, where voucher specimens are maintained.

Extraction and isolation. The stem bark (10.7 kg) was ground into powder and percolated with 95% ethanol. The dry extract (900 g) (F001) was partitioned between H₂O and CH₂Cl₂ to give a

H₂O layer (F002) and a CH₂Cl₂ layer. The residue of the CH₂Cl₂ layer (430 g) (F003) was partitioned between 90% MeOH and hexane, giving a MeOH layer (400 g) (F005) and a hexane layer (30 g) (F006). The MeOH layer (F005) was the most active fraction in the BST (LC₅₀ 1.02 μg/ml). Thus, a portion (190 g) of F005 was chromatographed over open silica gel columns directed by the BST test, using gradients of hexane-CHCl₃-MeOH. Collected fractions were combined into eight major pools (P1-P8) according to their TLC patterns. The bioactive P4 was repeatedly chromatographed over open silica gel columns followed by normal phase HPLC, 10% THF in MeOH-hexane (4-6)%, and reverse phase HPLC eluted with CH₃CN/H₂O (60/40 to 90/10) to give the colorless waxy compounds 1-3.

Pyranicin (1). White amorphous wax. (10 mg); $[α]_D^{23} = -9.7^0$ (c =0.008, CHCl₃); UV (MeOH) $λ_{max} = 216$ nm (log ε = 3.32); IR $ν_{max}$ cm⁻¹ (film on NaCl plate): 3418, 2928, 2854, 1748, 1456, 1319, 1086; CIMS (isobutane) m/z [MH]⁺ 597 (45), [MH-H₂O]⁺ 579 (100), [MH-2H₂O]⁺561 (93), [MH-3H₂O]⁺ 543 (43), [MH-4H₂O]⁺ 525 (3); EIMS diagnostic fragments see Figure 2; HRCIMS (isobutane) m/z 597.4711 for C₃₅H₆₅O₇ (calcd 597.4730); ¹H NMR (CDCl₃, 500 MHz) and ¹³C NMR (CDCl₃, 125 MHz) see Table 1.

Pyragonicin (2). White amorphous wax. (2 mg); $[α]_D^{23} = -25.6^{\circ}$ (c =0.008, CHCl₃); UV (MeOH) $λ_{max} = 215$ nm (log ε = 3.71); IR $ν_{max}$ cm⁻¹ (film on NaCl plate): 3479, 2920, 2851, 1748, 1456, 1318, 1084; CIMS (isobutane) m/z [MH]⁺ 597 (51), [MH-H₂O]⁺ 579 (100), [MH-2H₂O]⁺ 561 (54), [MH-3H₂O]⁺ 543 (13), [MH-4H₂O]⁺ 525 (1); EIMS diagnostic fragments see Figure 2; HRCIMS (isobutane) m/z 597.4741 for C₃₅H₆₅O₇ (calcd 597.4730); ¹H NMR (CDCl₃, 500 MHz) and ¹³C NMR (CDCl₃, 125 MHz) see Table 2.

Goniotrionin (3). A whitish wax (1.5 mg); UV (MeOH) $\lambda_{max} = 218$ nm (log $\epsilon = 3.46$); IR (film on NaCl plate) 3368, 2916, 2849, 1740, 1721, 1467, 1328, 1086, 1058, 841; CIMS (isobutane) m/z (%) [MH]⁺ 579 (1), [MH-H₂O]⁺ 561 (71), [MH-2H₂O]⁺ 543 (100), [MH-3H₂O]⁺ 525 (3); HRFABMS m/z 579.4597 for C₃₇H₆₈O₇ [MH]⁺ (calcd 579.4625); EIMS see Figure 4; ¹H and ¹³C NMR see Table 3.

Preparation of Mosher esters. To an acetogenin (0.5-1 mg, in 0.5 ml of CH_2Cl_2) were sequentially added pyridine (0.1 ml), 4-(dimethylamino)pyridine (0.1 mg), and 15 mg of (R)-(-)- α -methoxy- α -(trifluoromethyl)-phenylacetyl chloride. The mixture was stirred at rt from 4 hr to overnight, checked with TLC to make sure that the reaction was complete, and passed through a

disposable pipet (0.6 x 4 cm) containing silica gel (60-200 mesh) and eluted with 3 ml CH₂Cl₂. The CH₂Cl₂ residue, dried *in vacuo*, was redissolved in 1% NaHCO₃ (5 ml) and H₂O (2 x 5 ml); the CH₂Cl₂ layer was dried *in vacuo* to give the (S)-Mosher esters. Using (S)-(+)-α-methoxy-α-(trifluoromethyl)-phenylacetyl chloride gave the (R)-Mosher esters. Both yields were typically higher than 90%. For partial ¹H NMR assignments of 1a, 1b, 2a, and 2b see Tables 1 and 2. *Bioassays*. The bioactivities of extracts, fractions, and pure compounds were routinely assayed using a test for lethality to brine shrimp larvae (BST). The yellow fever mosquito larvae microtiter plate (YFM) assay was used to determine the relative pesticidal activities of compounds 1 and 2; rotenone was used as the positive pesticidal control standard. *In vitro* cytotoxicities, against six human tumor cell lines, were carried out at the Purdue Cancer Center, Cell Culture Laboratory, using standard 7-day MTT assays for A-549 (human lung carcinoma), MCF-7 (human breast carcinoma), HT-29 (human colon adenocarcinoma), A-498 (human kidney carcinoma), PC-3 (human prostate adenocarcinoma) and PACA-2 (human pancreatic carcinoma). Adriamycin is always used as a positive antitumor control in the same runs. Acknowledgments. This investigation was supported by R01 grant No. CA 30909 from the National Cancer Institute, National Institutes of Health. Stipend support for Feras Q. Alali was

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REFERENCES AND NOTES

- (a) Zeng, L.; Ye, Q.; Oberlies, N. H.; Shi, G.; Gu, Z.-M.; He, K.; McLaughlin, J. L. Nat. Prod. Rep. 1996, 13, 275-306.
 (b) Gu, Z.-M.; Zhao, G. X.; Oberlies, N. H.; Zeng, L.; McLaughlin, J. L. Recent Advances in Phytochemistry; Arnason, J. T.; Mata. R.; Romeo, J. T., Plenum Press: New York, 1995; Vol. 29, pp 249-310.
- 2. Shi, G.; Alfonso, D.; Fatope, M. O.; Zeng, L.; Gu, Z.-M.; Zhao, G.-X.; He, K.; MacDougal, J. M.; McLaughlin, J. L., J. Am. Chem. Soc. 1995, 117, 10409-10410.
- 3. Shi, G.; Kozlowski, J. F.; Schwedler, J. T.; Wood, K. V.; MacDougal, J. M.;
- 4. McLaughlin, J. L., J. Org. Chem. 1996, 61, 7988-7989.
- Alali, F.; Zeng, L.; Zhang, Y.; Ye, Q.; Hopp, D. C.; Schwedler, J.; McLaughlin, J. L. Bioorg. Med. Chem. 1997, 5, 549-555.
- (a) McLaughlin, J. L., Methods in Plant Biochemistry, Hostettmann, K., Academic Press: London, 1991, Vol. 6, 1-35.
 (b) Meyer, B. N.; Ferrigni, N. R.; Putnam, J. E.; Jacobson, L. B.; Nichols, D. E.; McLaughlin, J. L., Planta Med. 1982, 45, 31-34.

- 7. Born, L.; Lieb, F. J.; Lorentzen, P.; Moeschler, H.; Nonfon, M.; Söllner, R.; Wendisch, D. Planta Med. 1990, 56, 312-316.
- (a) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H., J. Am. Chem. Soc. 1991, 113, 4092-4096.
 (b) Rieser, M. J.; Fang, X. P.; Anderson, J. E.; Miesbauer, L. R.; Smith, D. L.; McLaughlin, J. L. Helv. Chim. Acta 1993, 76, 2433-2444; erratum Helv. Chim. Acta 1994, 77, 882.
 (c) Rieser, M. J.; Hui, Y.-H.; Rupprecht, J. K.; Kozlowski, J. F.; Wood, K. V.; McLaughlin, J. L.; Hanson, P. R.; Zhuang, A.; Hoye, T. R. J. Am. Chem. Soc. 1992, 144, 10203-10213.
- 9. Hoye, T. R.; Hanson, P. R.; Hansenwinkel, L. E.; Ramirez, E. A.; Zhuang, Z. Tetrahedron Lett. 1994, 35, 8529-8532.
- 10. Ye, Q.; Alfonso, D.; Evert, D.; McLaughlin, J. L. Bioorg. Med. Chem. 1996, 4, 537-545.
- 11. Rieser, M. J., Ph.D., Annonaceous Acetogenins from the Seeds of *Annona muricata*. Ph.D. Thesis, Purdue University, West Lafayette, 1993.
- 12. Zeng, L.; Zhang, Y.; McLaughlin, J. L. Tetrahedron Lett. 1996, 37, 5449-5452.
- 13. Fujimoto, Y.; Murasaki, C.; Shimada, H.; Nishioka, S.; Kakinuma, K.; Singh, S.; Singh, M.; Gupta, Y. K.; Sahai, M. Chemical and Pharmaceutical Bulletin, 1994, 42, 1175-1184.
- 14. Tam, V. T.; Chaboche, C.; Figadere, B.; Chappe, B.; Hieu, B. C.; Cave, A. Tetrahedron Lett. 1994, 35, 883-886.
- Wu, F.-E.; Zeng, L.; Zhao, G.-X.; Zhang, Y.; Schwedler, J. T.; McLaughlin, J. L. J. Nat. Prod. 1995, 58, 902-908.
- Wu, F.-E.; Zeng, L.; Zhao, G.-X.; Zhang, Y.; Schwedler, J. T.; McLaughlin, J. L. J. Nat. Prod. 1995, 58, 909-915.
- 17. Anonymous, World Health Organization Report Series, 1970, 443, 66.
- 18. The 7-day MTT in vitro tests against human cell lines followed the standard protocols as previously described.⁴
- 19. Oberlies, N. H.; Chang, C.-J.; McLaughlin, J. L. J. Med. Chem. 1997, 40, 2102-2106.
- 20. Ahammadsahib, K. I.; Hollingworth, R. M.; Hui, Y.-H.; McLaughlin, J. L. Life Sci. 1993, 53, 1113-1120.
- 21. Morre, J. D.; Decabo, R.; Farley, C.; Oberlies, N. H.; McLaughlin, J. L. Life Sci. 1995, 56, 343-348.
- 22. Wolvetang, E. J.; Johnson, K. L.; Krauer, K.; Ralph, S. J.; Linnane, A. W. FEBS Lett. 1994, 339, 40-44.
- 23. Oberlies, N. H.; Croy, V. L.; Harrison, M. L.; McLaughlin, J. L. Cancer Lett. 1997, 115, 73-79.
- 24. Alali, F. Q.; Kaakeh, W.; Bennett, G. W.; McLaughlin, J. L. 1998, J. Econ. Entomol., (accepted).